

An In-situ Method for Organic Aerosol Speciation via Thermal Desorption Gas Chromatography – Mass Spectrometry

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Objective

Automated, hourly measurements of organic marker compounds in ambient aerosols.

Science Questions

What are the major sources for organic aerosols?

How important is secondary organic aerosol (SOA)?

What compounds comprise SOA, and how are they formed?

Organic Matter

(blue)

20% – 40%

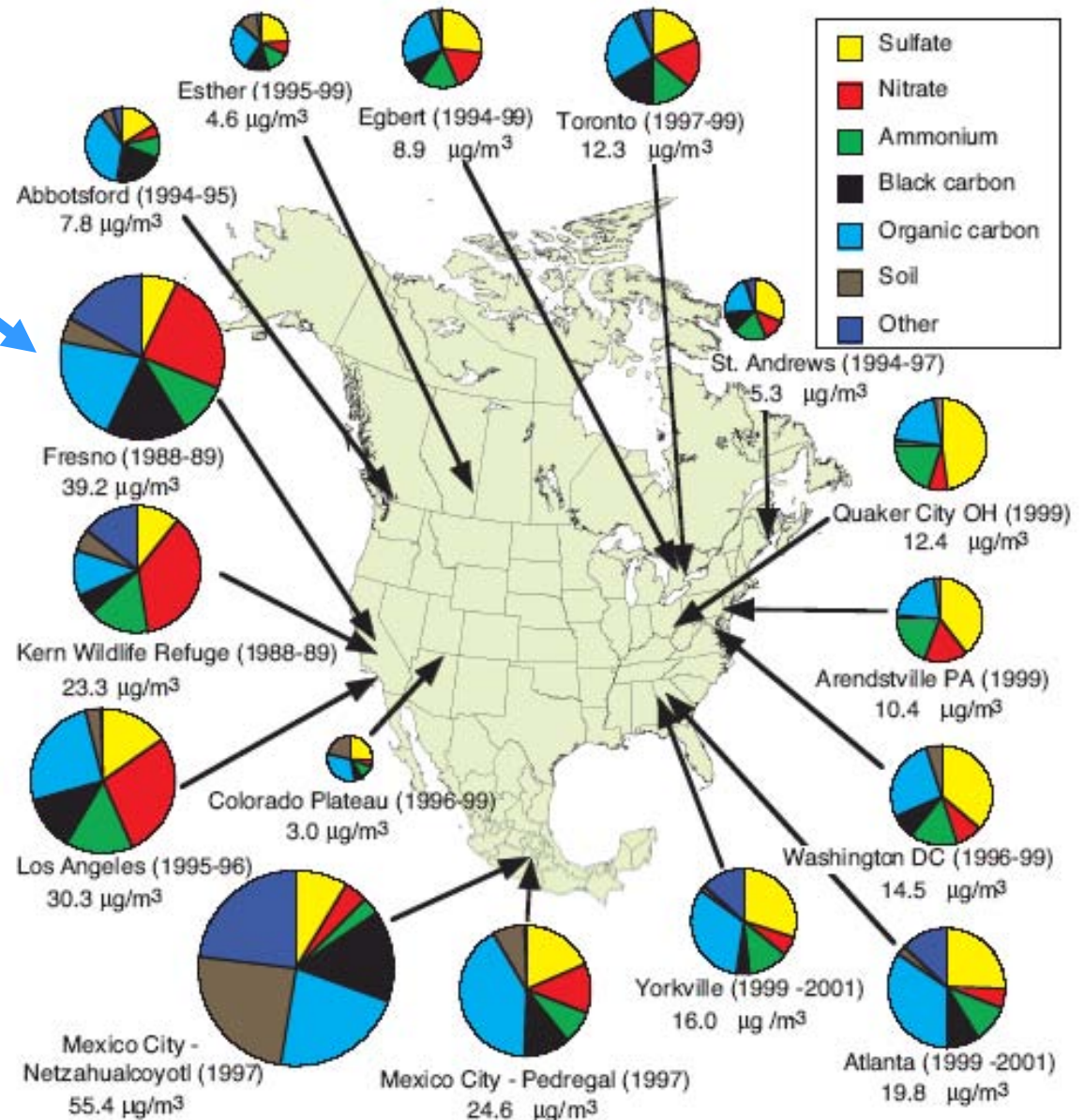
of $PM_{2.5}$

NARSTO
Assessment on
Particulate Matter

February 2003

Chapter 6

(C. Blanchard)



What are the Constituents of Organic PM?

Hundreds – Thousands of individual compounds:

- Alkanes
- Polycyclic aromatic hydrocarbons
- Substituted phenols
- Mono- and Di-carboxylic acids
- Sugars

Analysis of Organic Matter at the Compound Level

B. Simoneit, M. Mazurek, G. Cass
J. Schauer, W. Rogge, M. Fraser

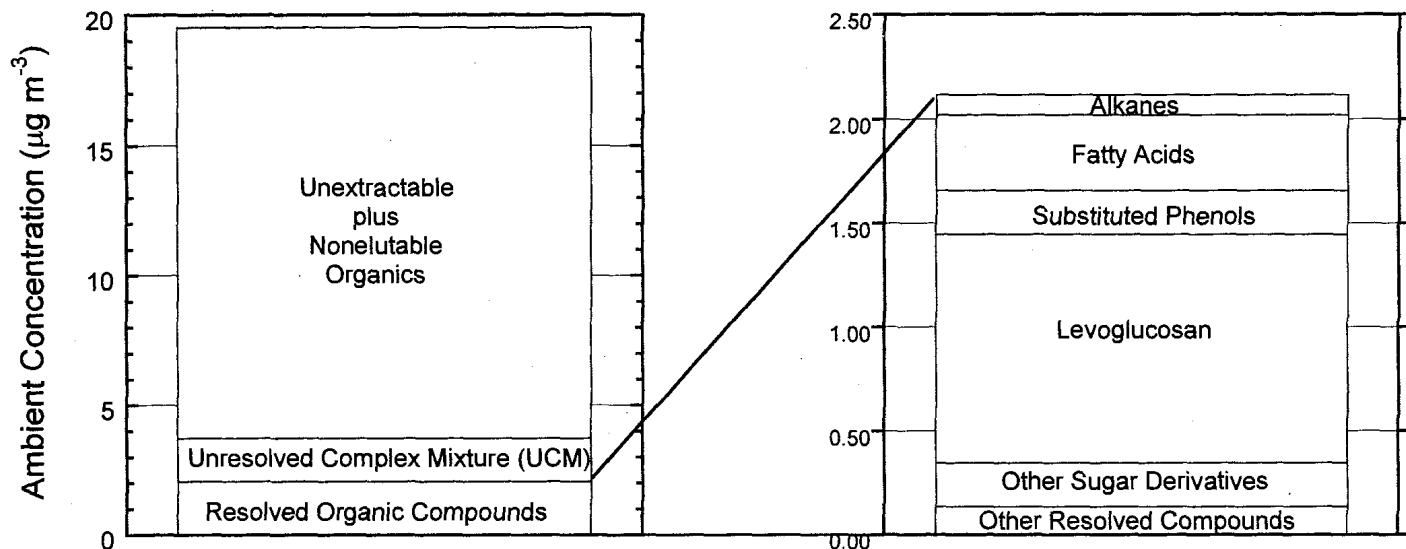


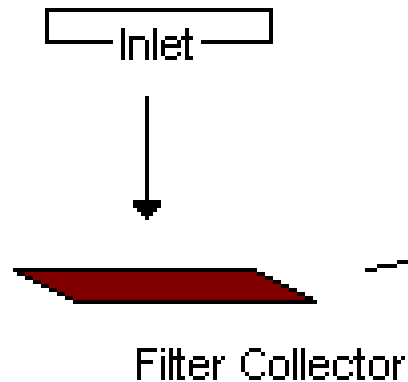
FIGURE 1. Mass balance on the fine particle organic compounds measured at Bakersfield, CA, between December 26 and 28, 1999.

Schauer and Cass, Environ Sci Technol 43: 1826, 2000

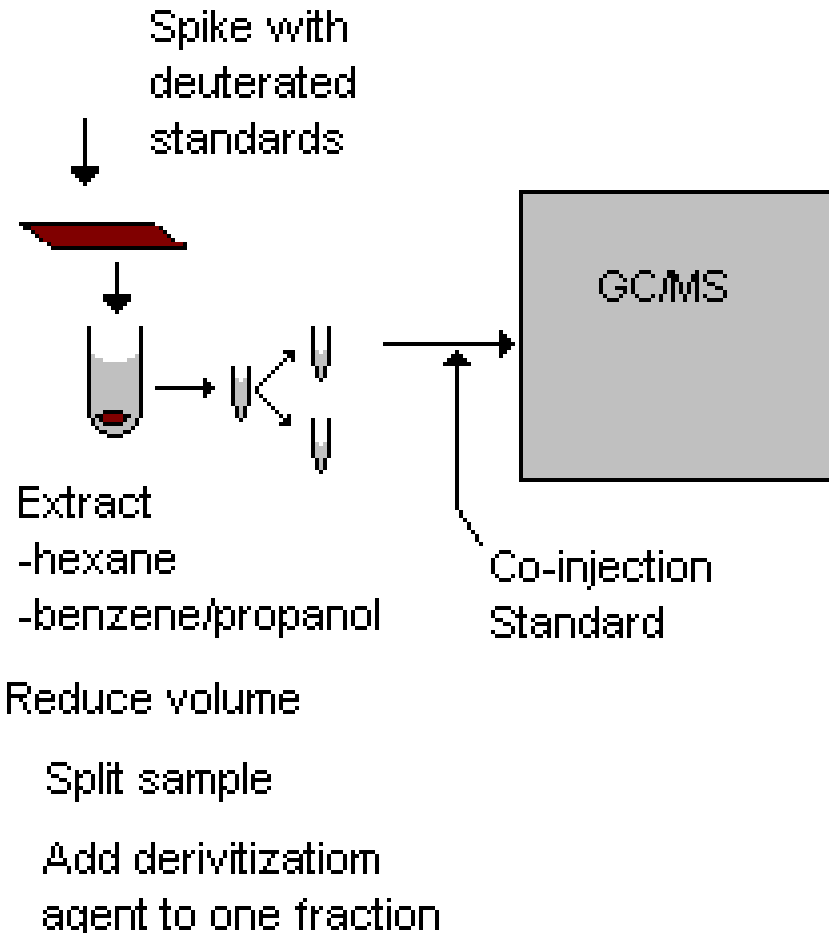
Resolved compounds ~10% of total organic matter.
Analysis provides marker compounds, such as levoglucosan,
useful for tracing sources.

Steps in GC/MS Analysis of Filter Samples

Sample Collection



Laboratory Analysis



Identification at
compound level:

-- relate to sources,
--identify pathways

**BUT: Tedious, yields
limited time resolution**

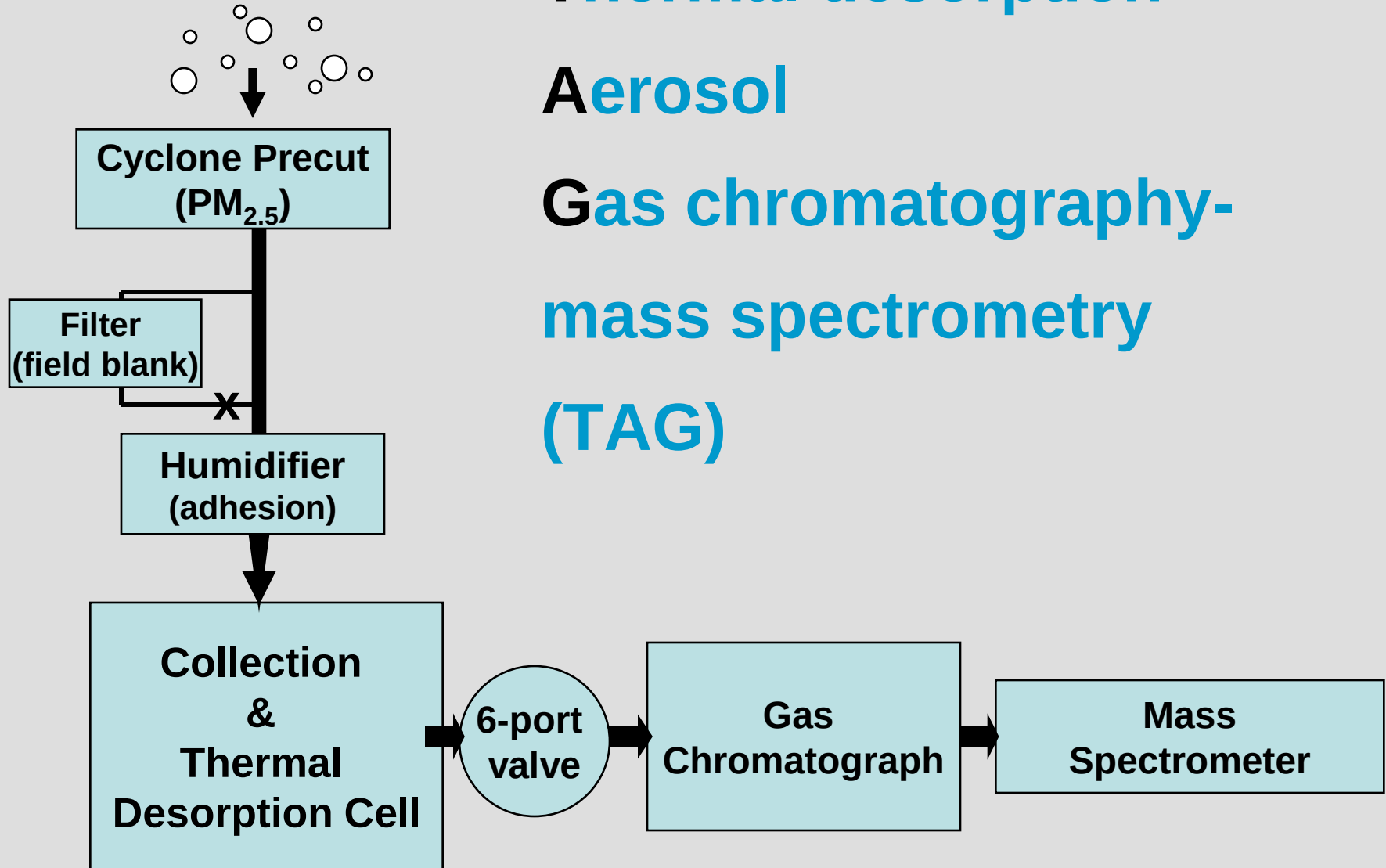
Automated Measurements:

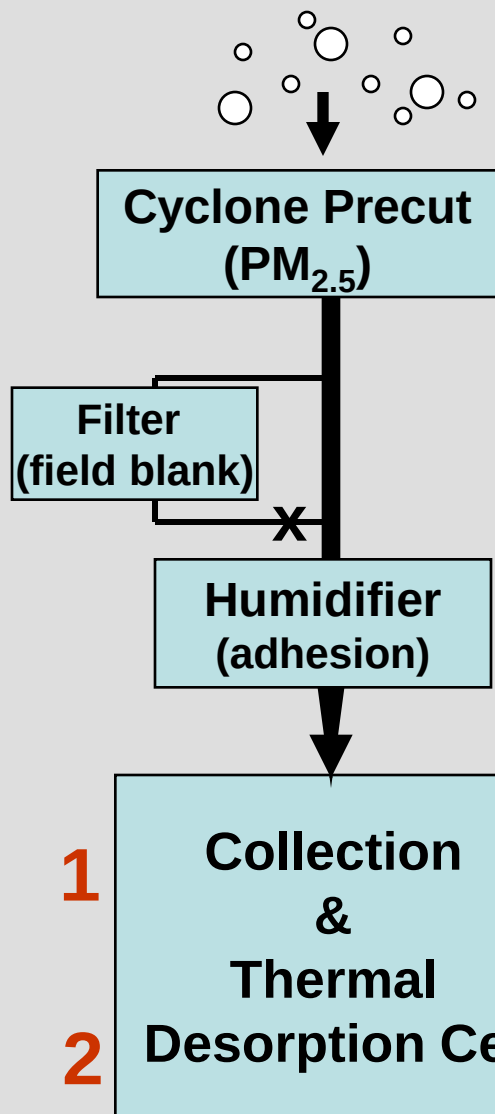
In-Situ Mass Spectrometry (ATOFMS, AMS, others) –
high time resolution
no individual compound identification

**Our Aim – to combine advantages of
the speciation of filter measurements with
the time resolution of automated methods :**

Separating / Identifying / Quantifying individual organic
aerosol marker compounds with hourly time resolution

Thermal desorption Aerosol Gas chromatography- mass spectrometry (TAG)





1. Aerosol Collection:

- Humidification / Inertial Impaction (30 C)

2. Sample transfer:

- Thermal Desorption (50-300 C)

3. Chemical separation:

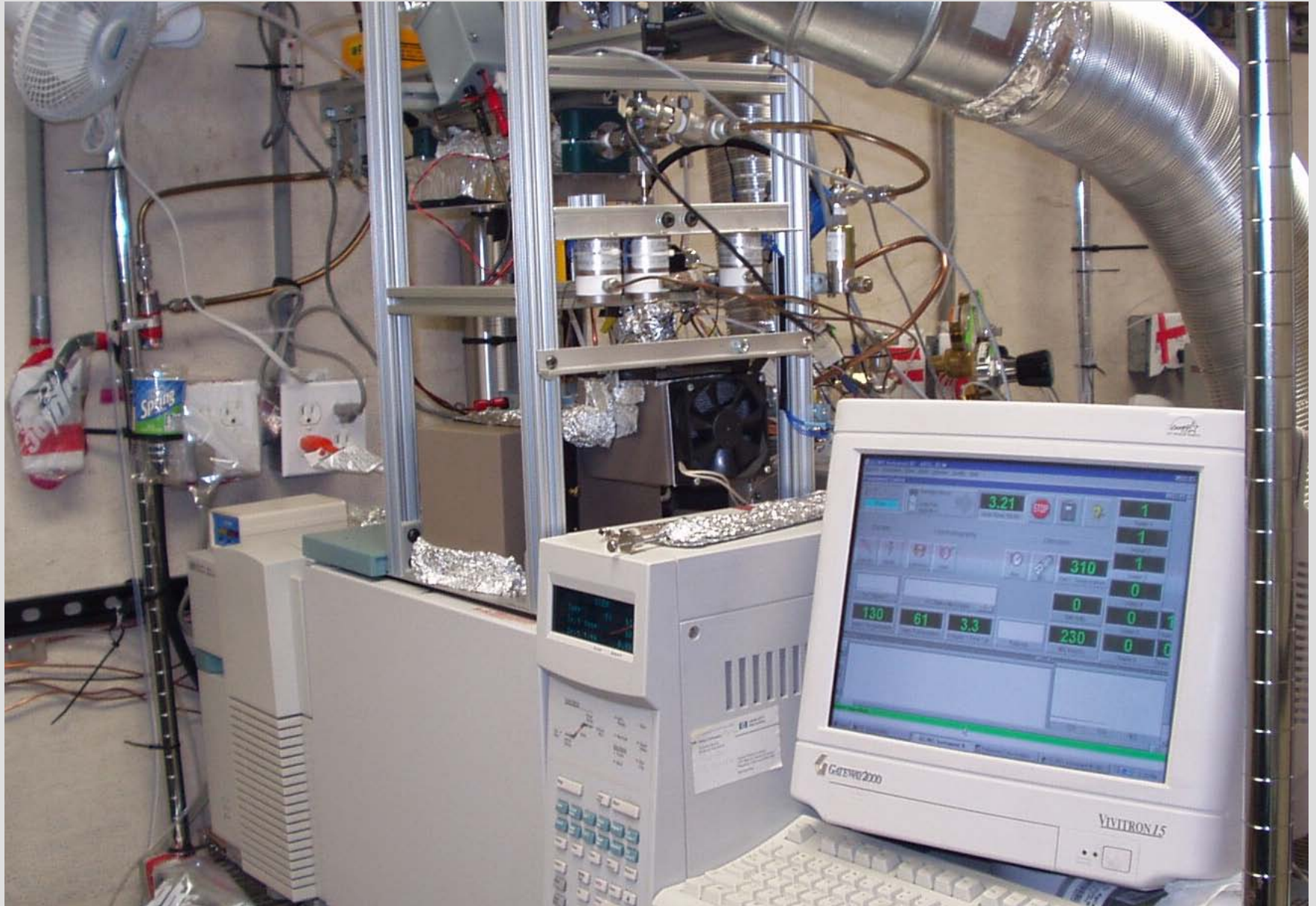
- Gas Chromatography

4. Identification & quantification:

- Mass Spectrometry (and/or FID)
- Spectrum + Retention Time Matching
- Standards

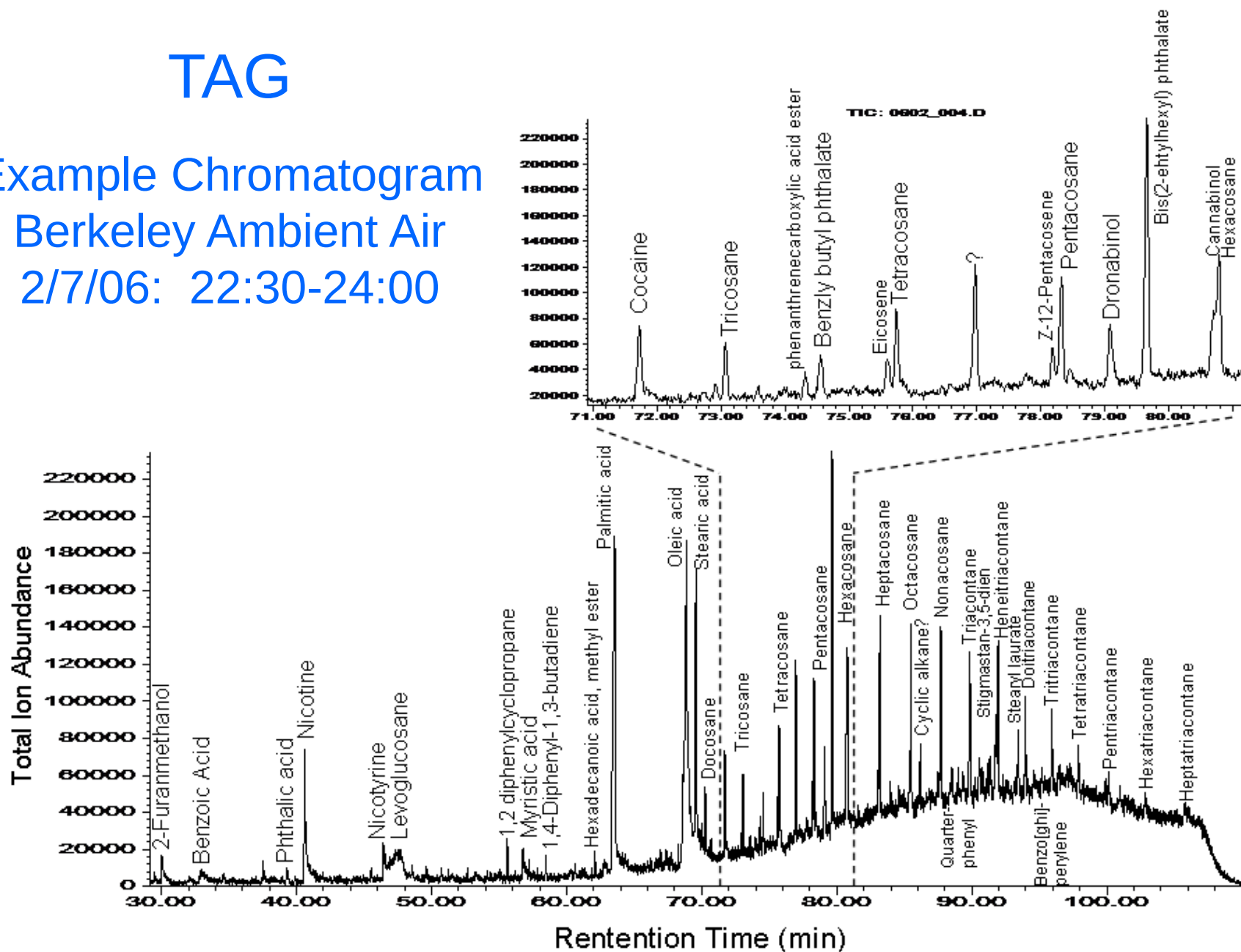
Cycle time = 1 hour,
Provision for standards, automated blanks

TAG in Nova Scotia



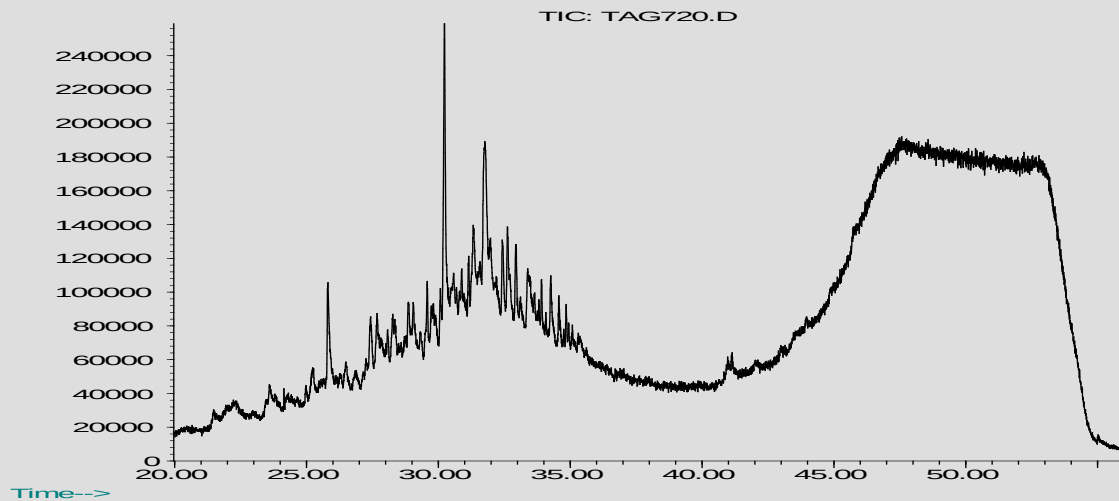
TAG

Example Chromatogram Berkeley Ambient Air 2/7/06: 22:30-24:00



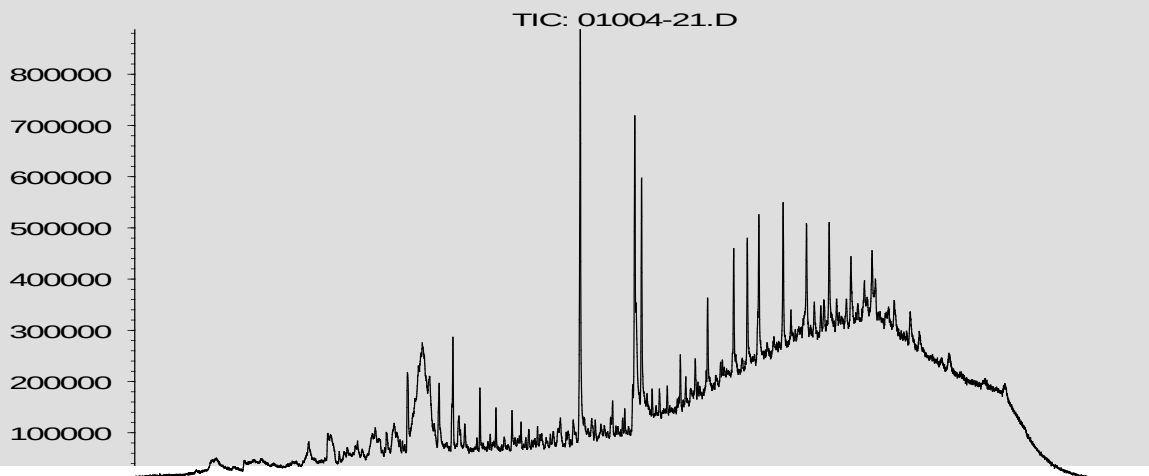
Comparison of Chromatograms from Nova Scotia (remote site) with Berkeley, CA (urban site)

Abundance



Chebogue Point,
Nova Scotia 8/7/04
-- more oxygenated
-- elute sooner

Abundance



Berkeley, CA
1/10/04

Some Identified Compounds from Nova Scotia

5-Hexenoic acid, 5-methyl-	$C_7H_{12}\underline{O}_2$
5,6-Dihydropyran-2-one, 5-acetoxy-6-(1,2-epoxypropyl)-	$C_{10}H_{12}\underline{O}_5$
Pelletierine	$C_8H_{15}\underline{NO}$
2,2-Dimethyl-3-heptene trans	C_9H_{18}
4-Pentenoic acid, 2-acetyl-2,3-dimethyl-,ethyl ester	$C_{11}H_{18}\underline{O}_3$
4s,6s-Dimethyl-7R-acetoxy-3-nonanone (acetyl serricornin)	$C_{13}H_{24}\underline{O}_3$
2(3H)-Furanone, 5-methyl-	$C_5H_6\underline{O}_2$
Spiro[1,3-dioxolane-2,2'-[6,7]diazabicyclo[3.2.2]non-6-ene	$C_9H_{14}N_2\underline{O}_2$
2(3H)-Furanone, 3-acetyldihydro-	$C_6H_8\underline{O}_3$
1,6-Dioxaspiro[4,4]nonane-2,7-dione	$C_7H_8\underline{O}_4$
Cyclohexanone, 2-ethyl-	$C_8H_{14}\underline{O}$
2,3-Pinandediol	$C_{10}H_{18}\underline{O}_2$
Phthalic acid	$C_8H_6\underline{O}_4$
3,5-di-tert-Butyl-4-hydroxybenzaldehyde	$C_{15}H_{22}\underline{O}_2$
Dibenz[c,d]hydrooxepin	$C_{14}H_{12}\underline{O}$

Timelines for Individual Compounds

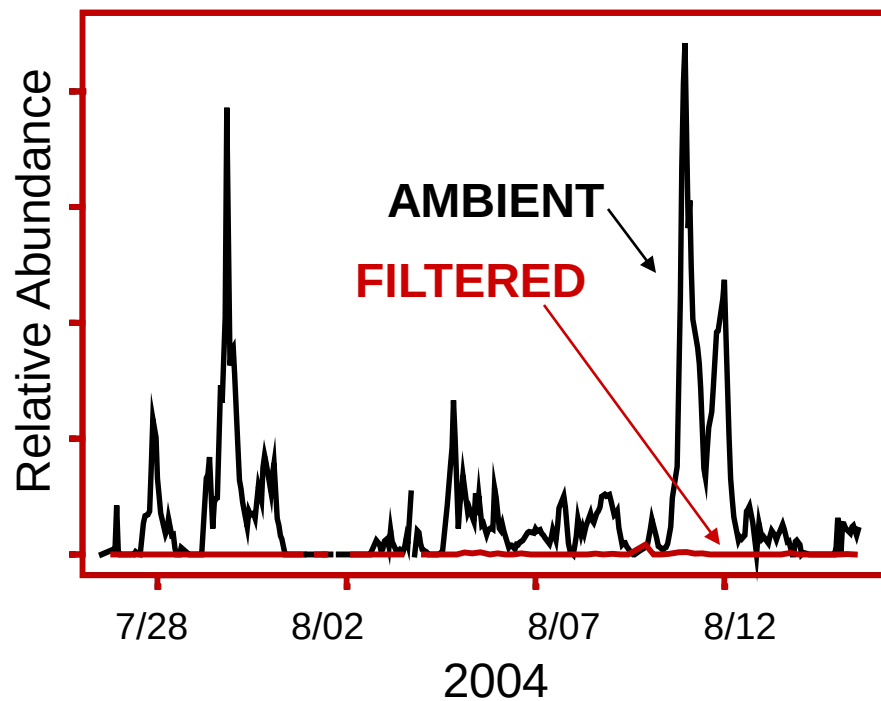
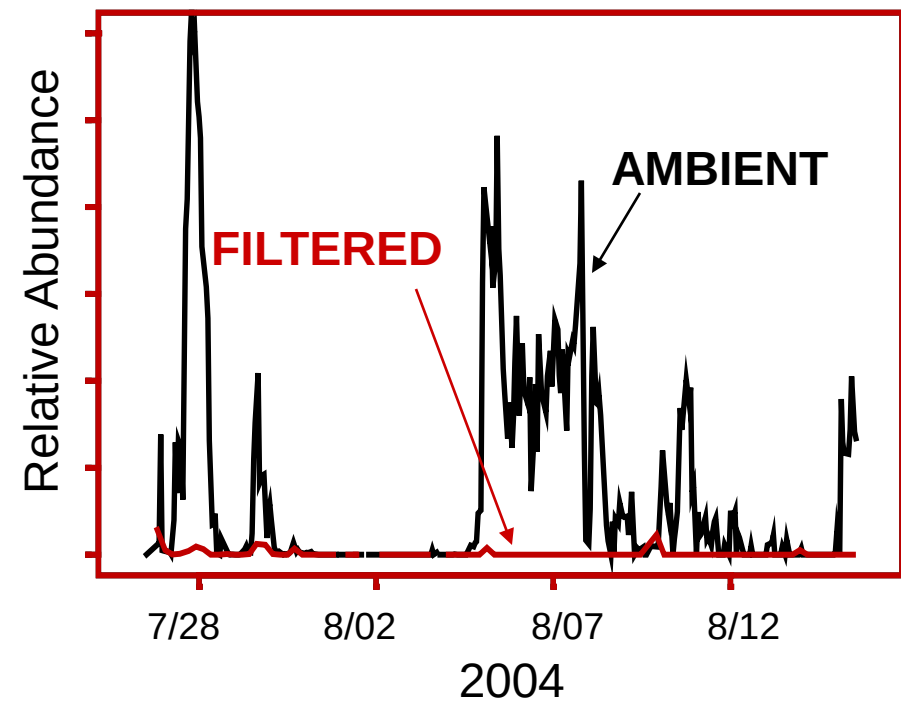
Chebogue Point Nova Scotia 2004

Compound A:

4-Pentenoic acid, 2-acetyl-2,3-dimethyl-,ethyl ester ($C_{11}H_{18}O_3$)

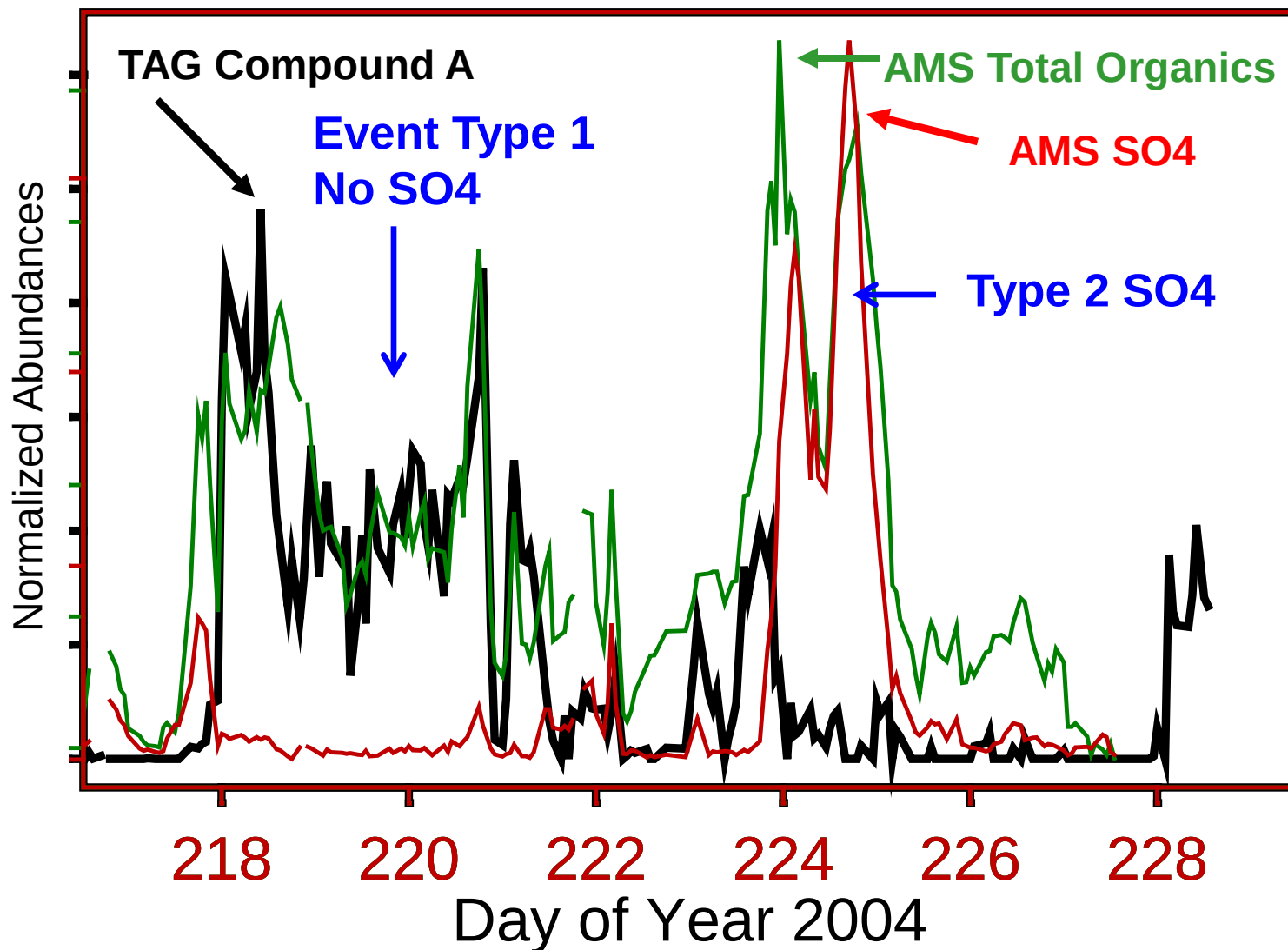
Compound B:

1,6-Dioxaspiro[4,4]nonane-2,7-dione ($C_7H_8O_4$)



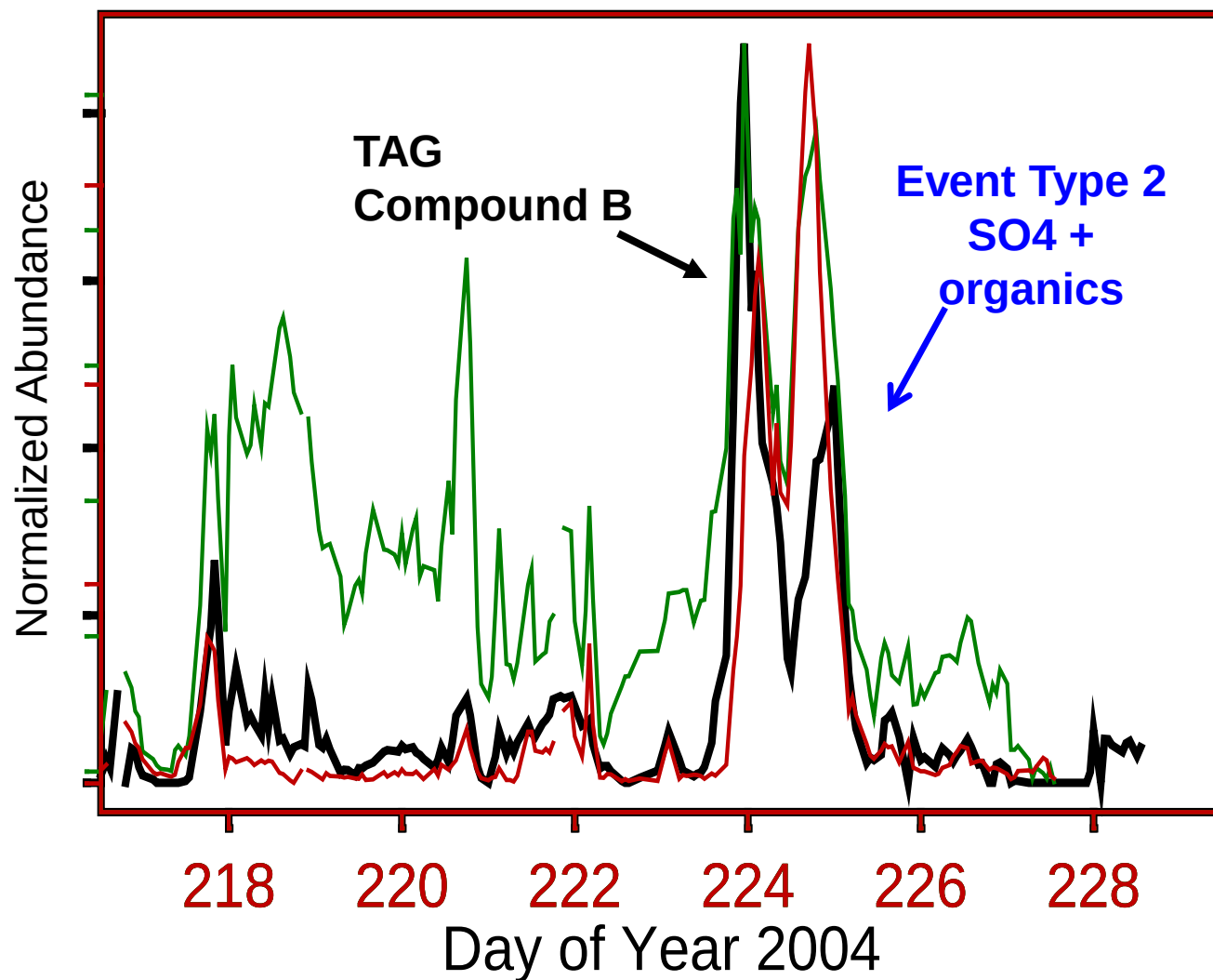
TAG Compound A vs AMS Total Organics and Total SO4

4-Pentenoic acid, 2-acetyl-2,3-dimethyl-,ethyl ester ($C_{11}H_{18}O_3$)



Tag Compound B vs AMS Organics and Sulfates

1,6-Dioxaspiro[4,4]nonane-2,7-dione (C₇H₈O₄)

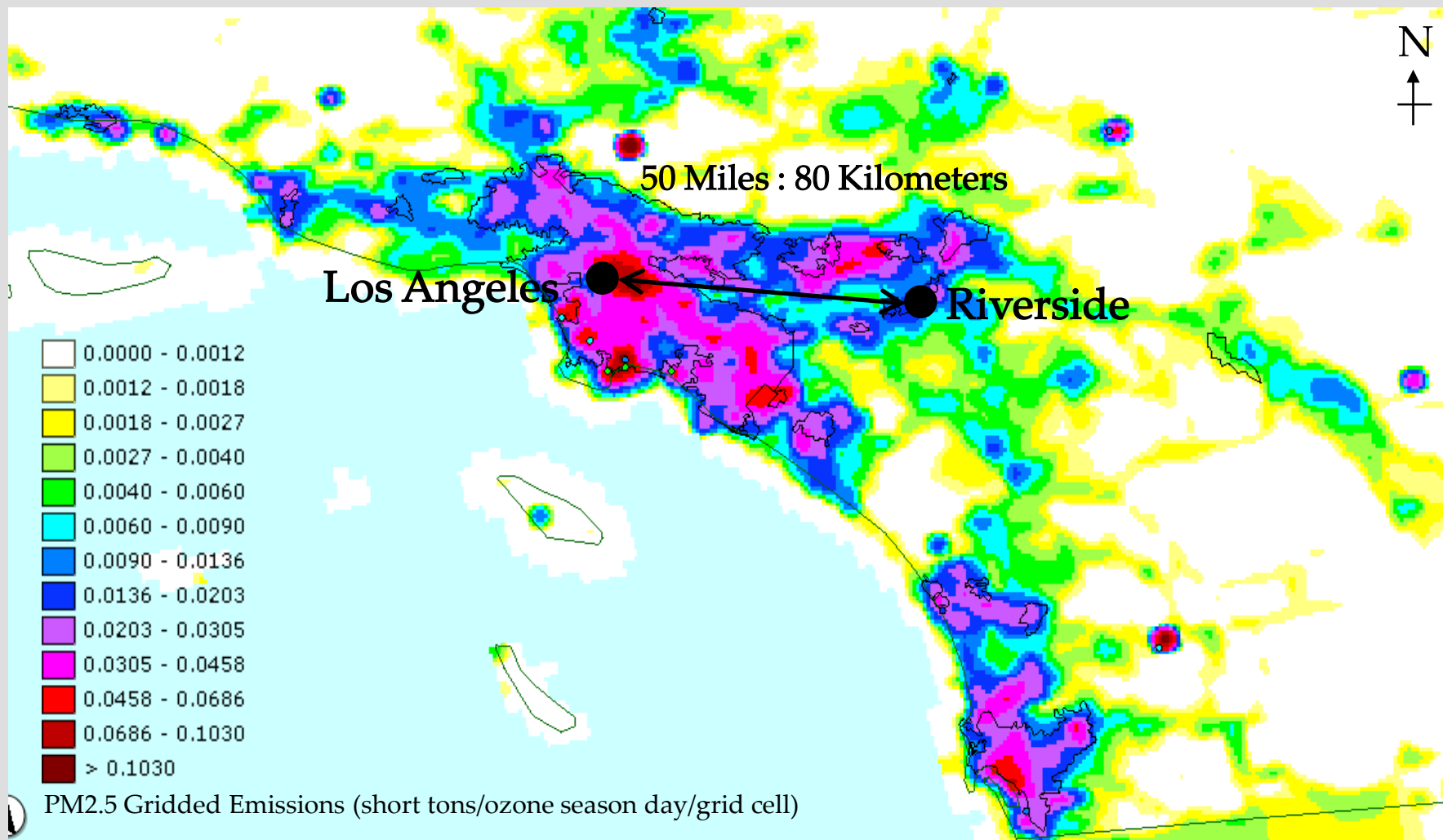


These two compounds represent two different events.

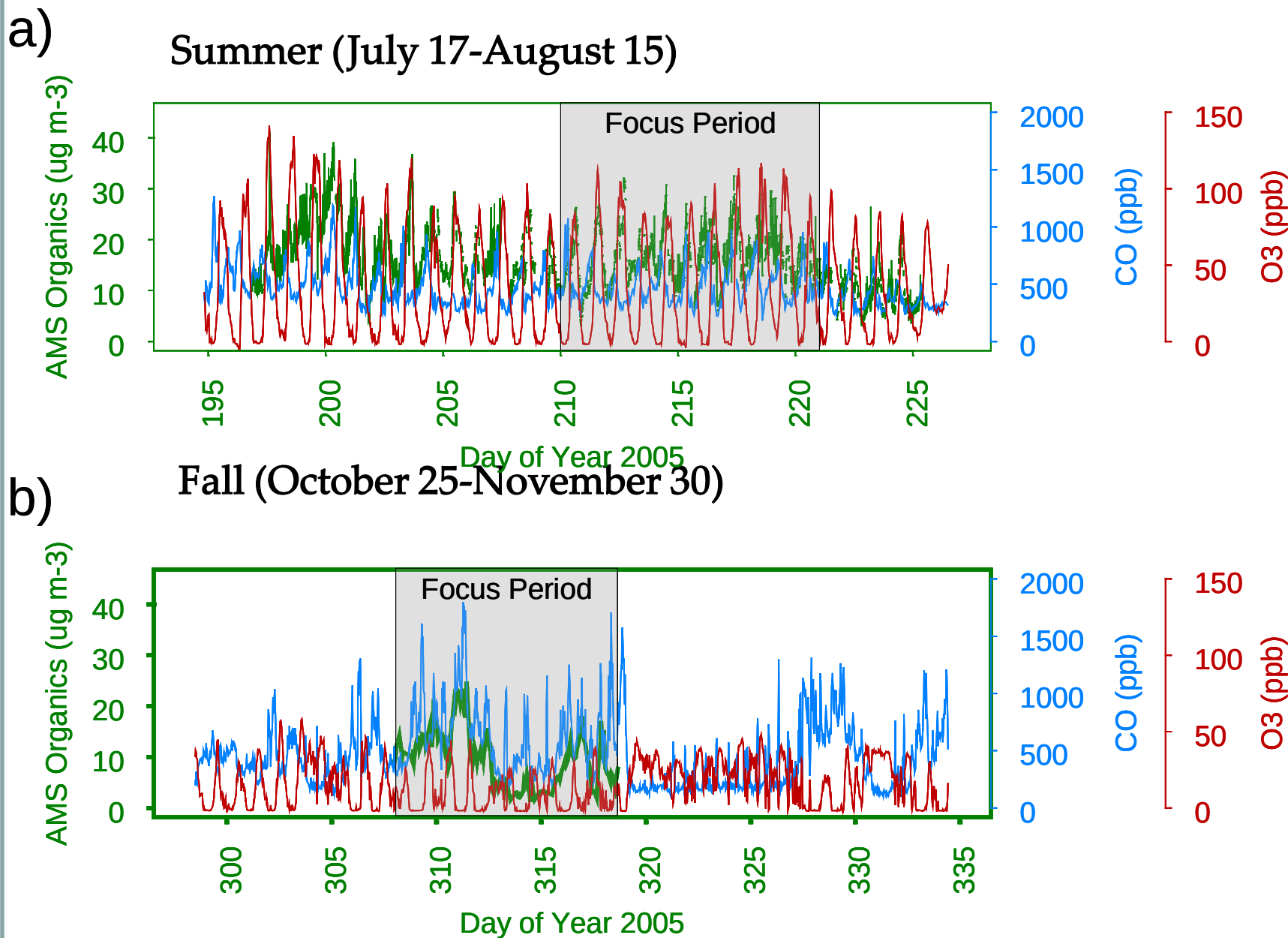
AMS data (Aerodyne Research, Inc.: Worsnop et al.)

Study of Organic Aerosols at Riverside (SOAR)

Location: University of California, Riverside



SOAR Measurement Periods

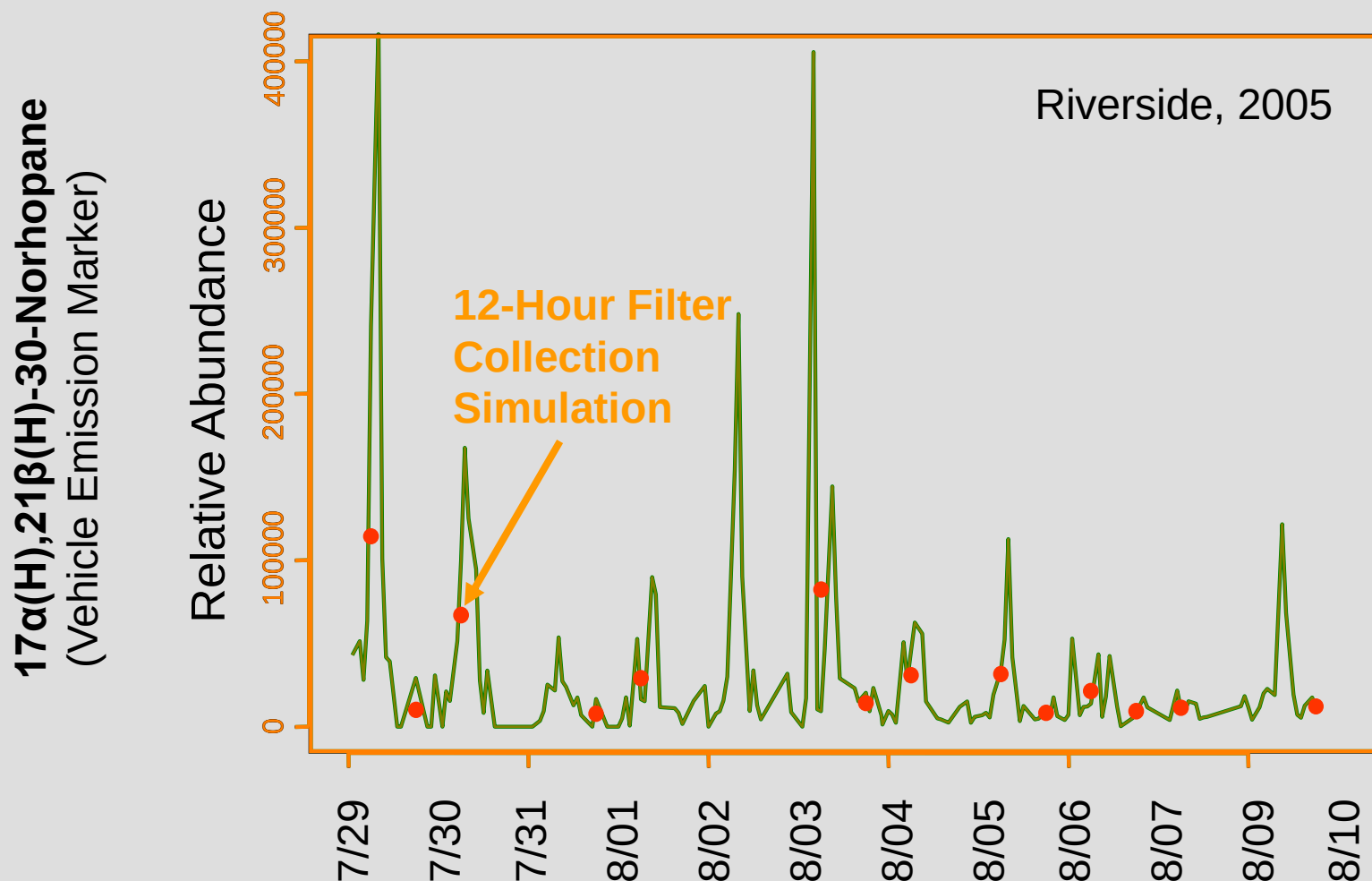


Very different Photochemistry (and Meteorology)

SOAR TAG Measurements and Source Apportionment

- Resolved >300 organic marker compounds hourly
- Positive Matrix Factorization (PMF)
 - 124 of 300 resolved TAG organic compounds in Summer
 - 141 of 300 resolved TAG organic compounds in Fall
- Identified organic aerosol sources based on PMF groupings
 - 9 in Summer
 - 6 in Fall

Hourly (TAG) data captures dynamics missed by 12 hour filter collections

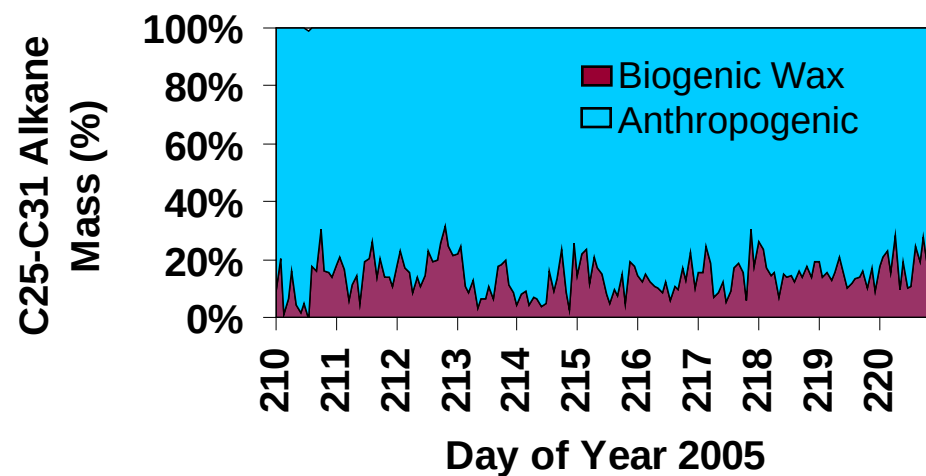


Biogenic vs. Anthropogenic Alkanes

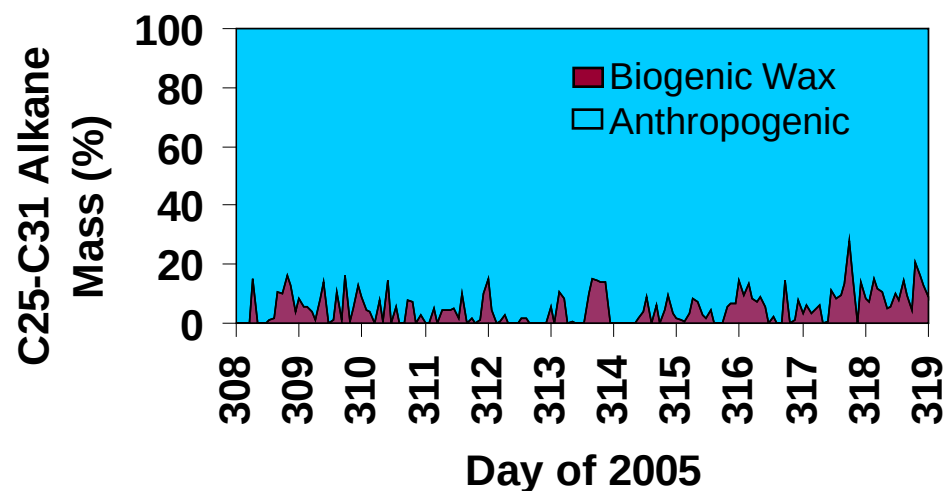
$$C_{wax} = \sum_{n=25}^{31} \left[C_n - \frac{(C_{n-1} + C_{n+1})}{2} \right]$$

Adapted from Simoneit 1989
J. Atmos. Chem.

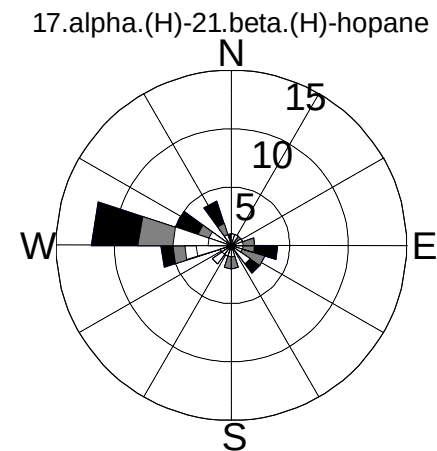
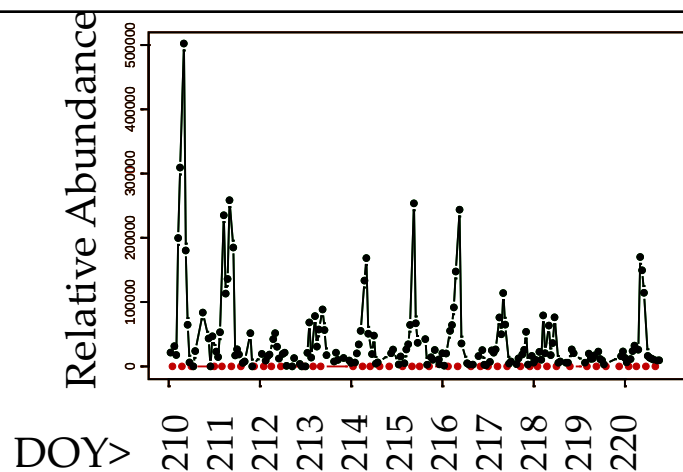
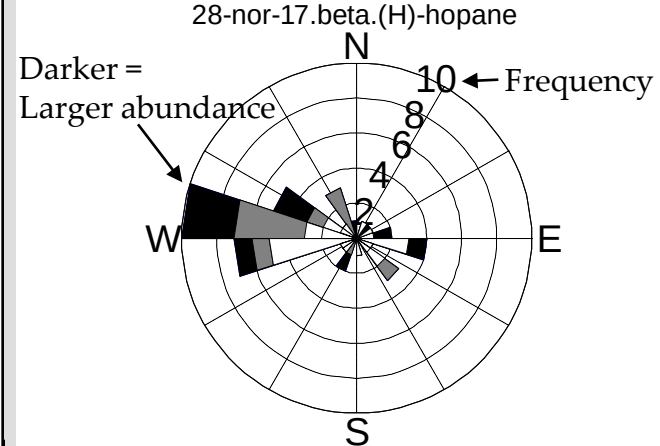
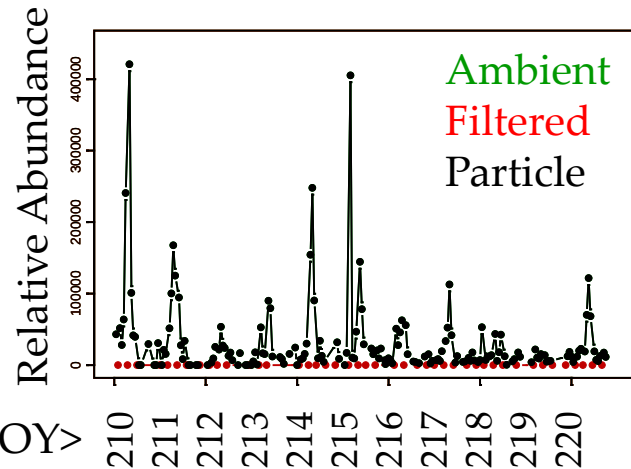
SUMMER



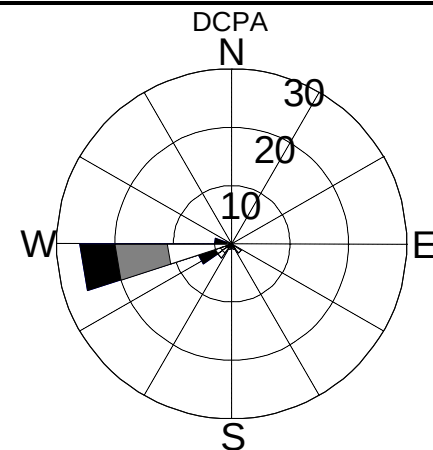
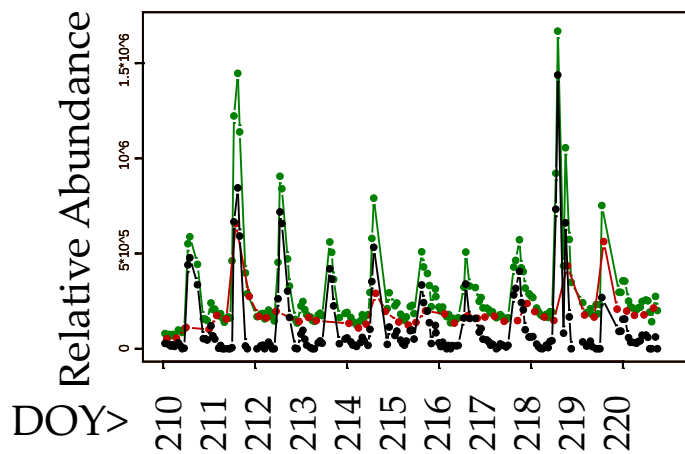
FALL



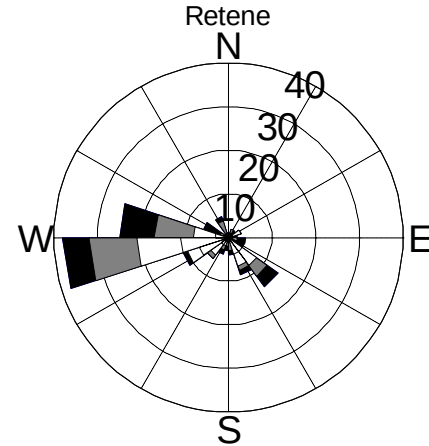
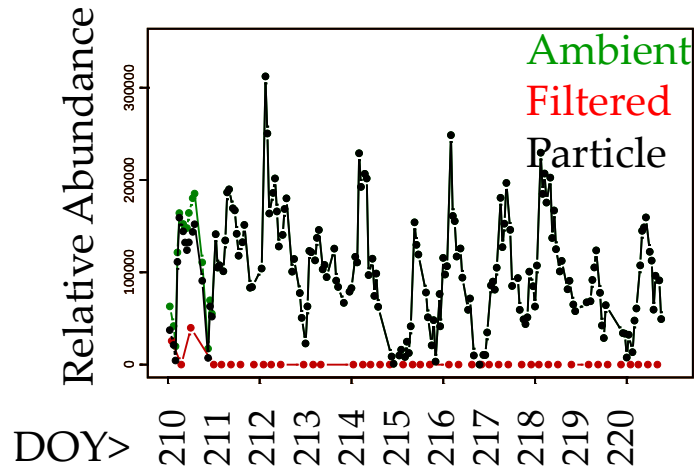
Primary
Anthropogenic
(vehicle)



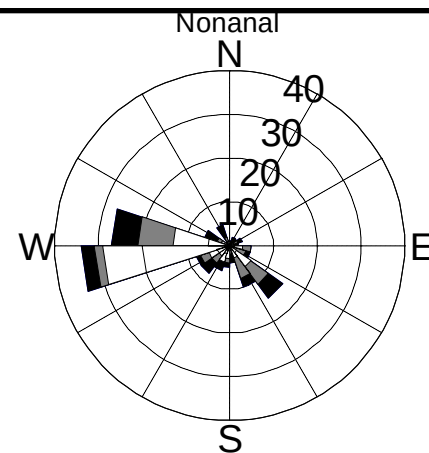
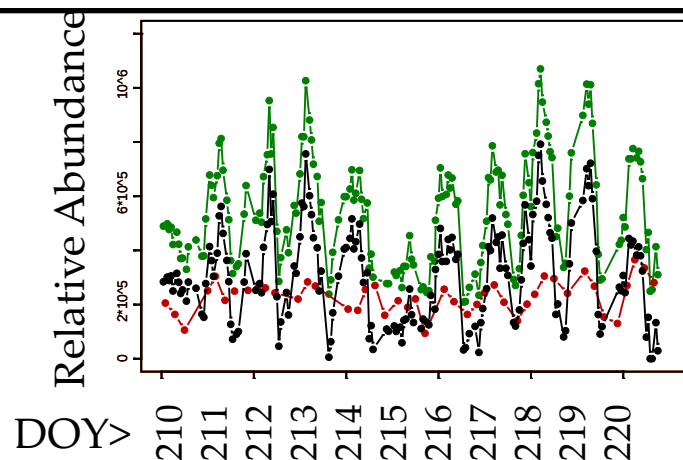
Herbicide



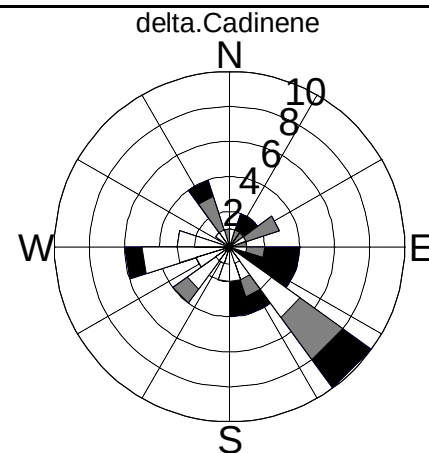
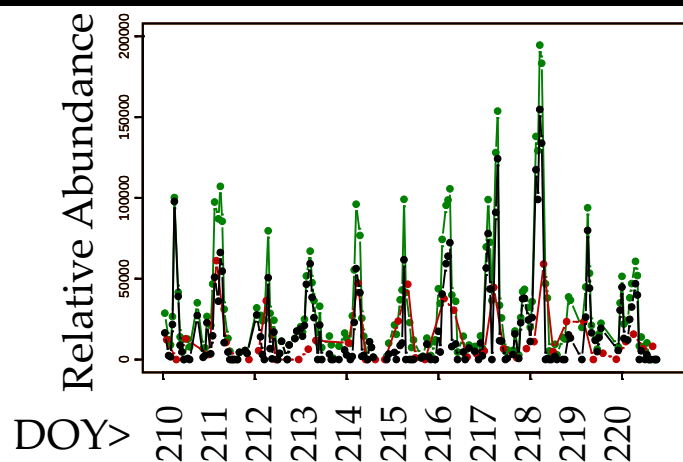
Biomass Burning



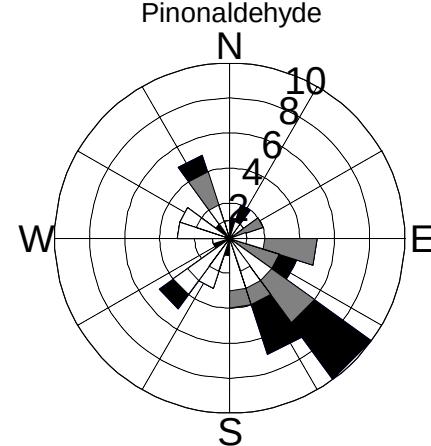
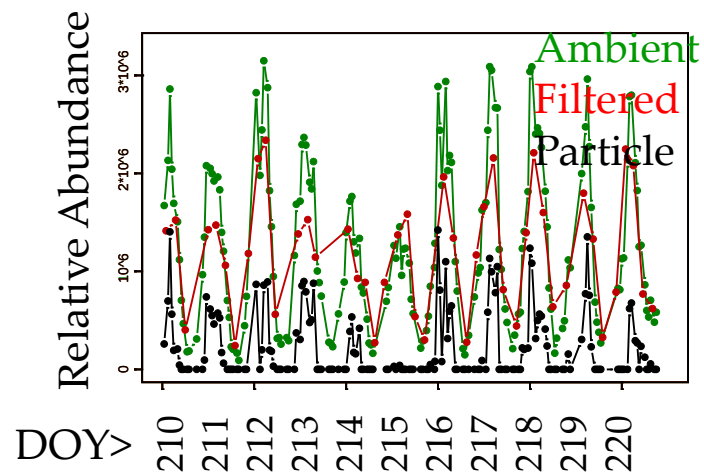
Meat Cooking



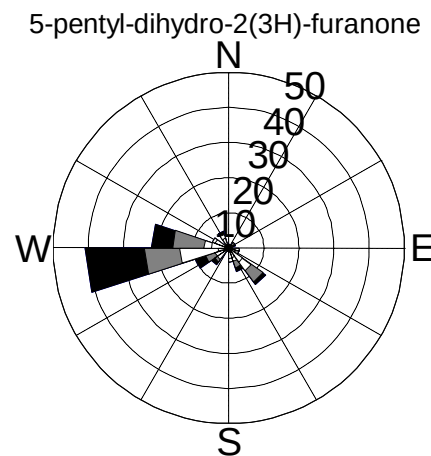
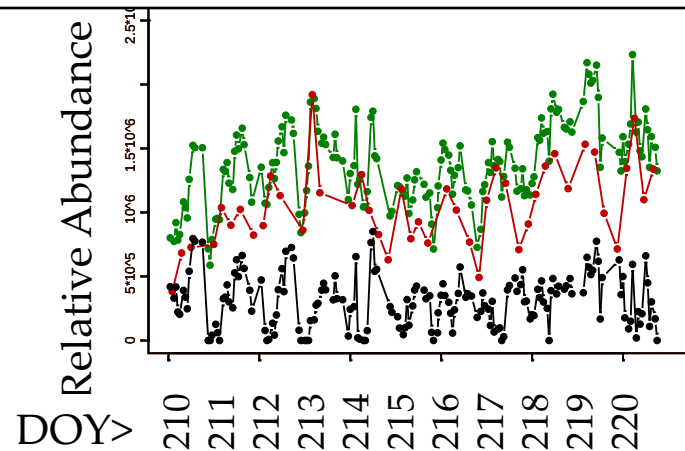
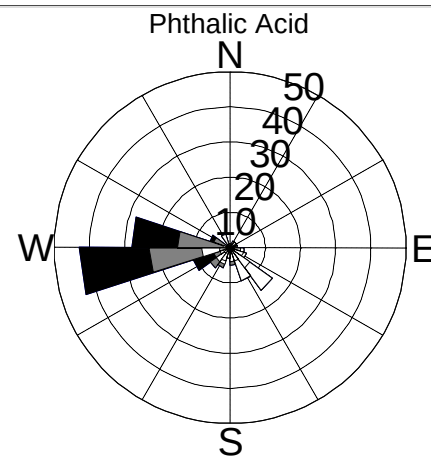
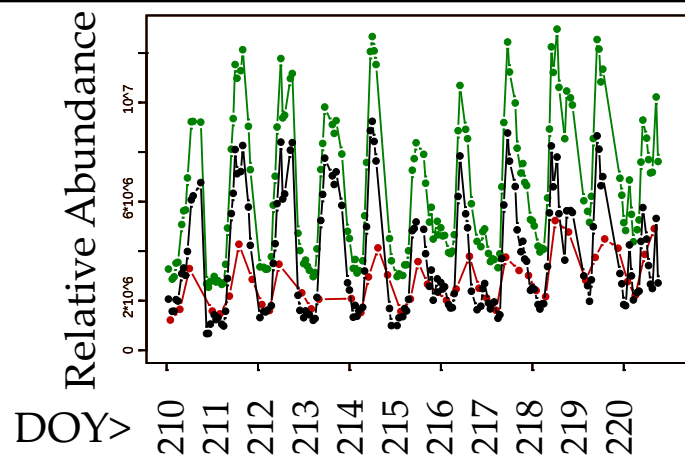
Primary Biogenic



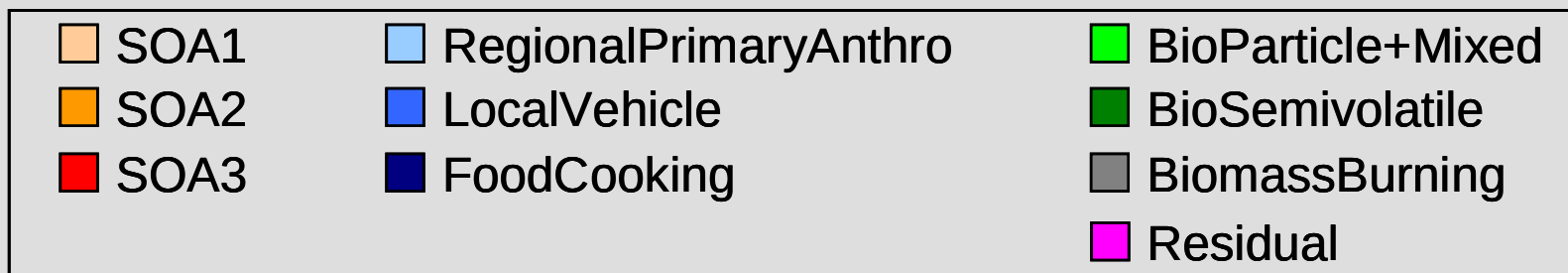
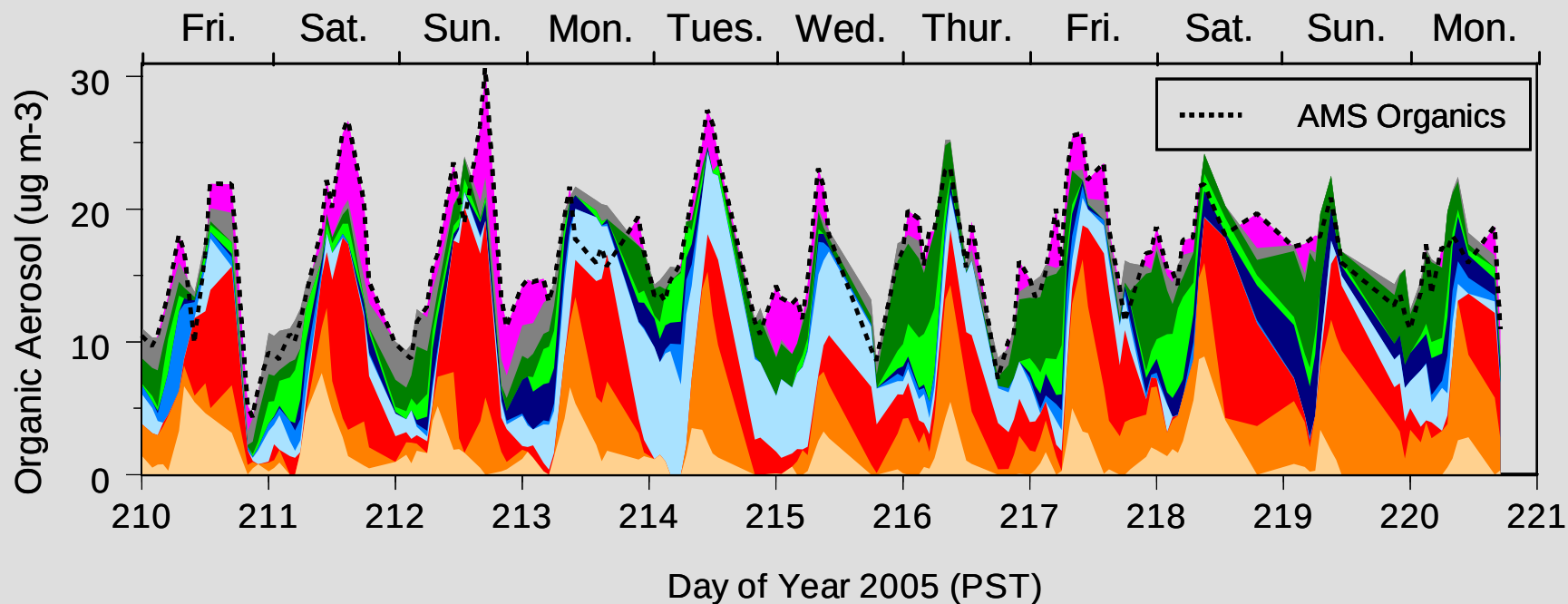
Secondary Biogenic



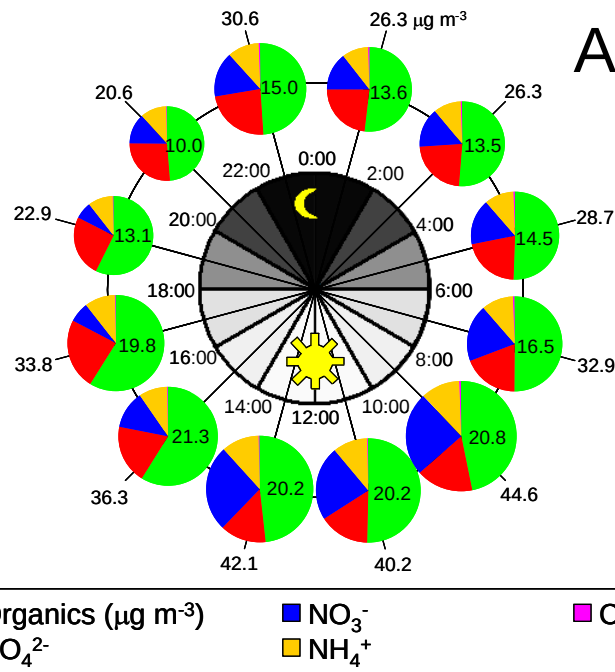
Secondary Anthropogenic



SOAR Summer: PMF Results

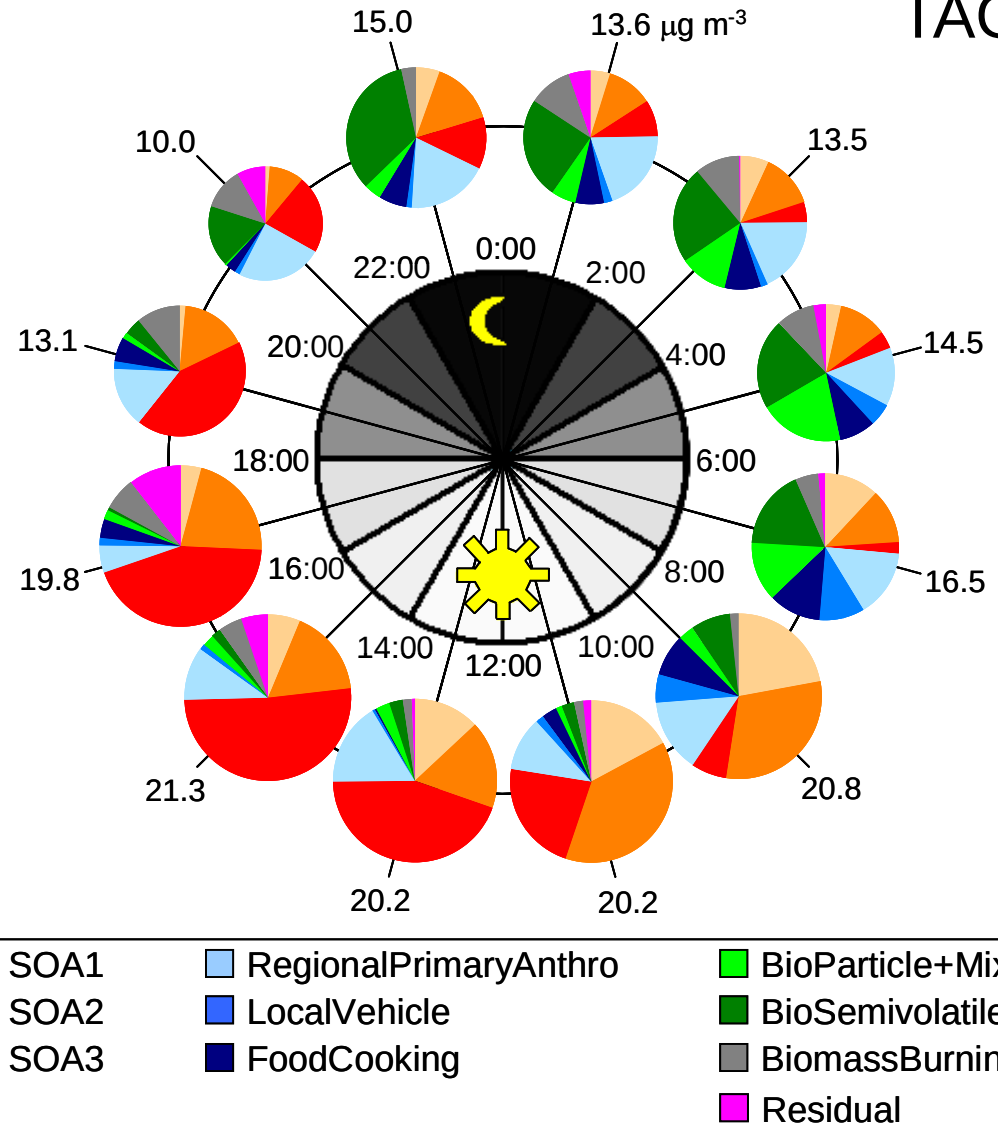


AMS



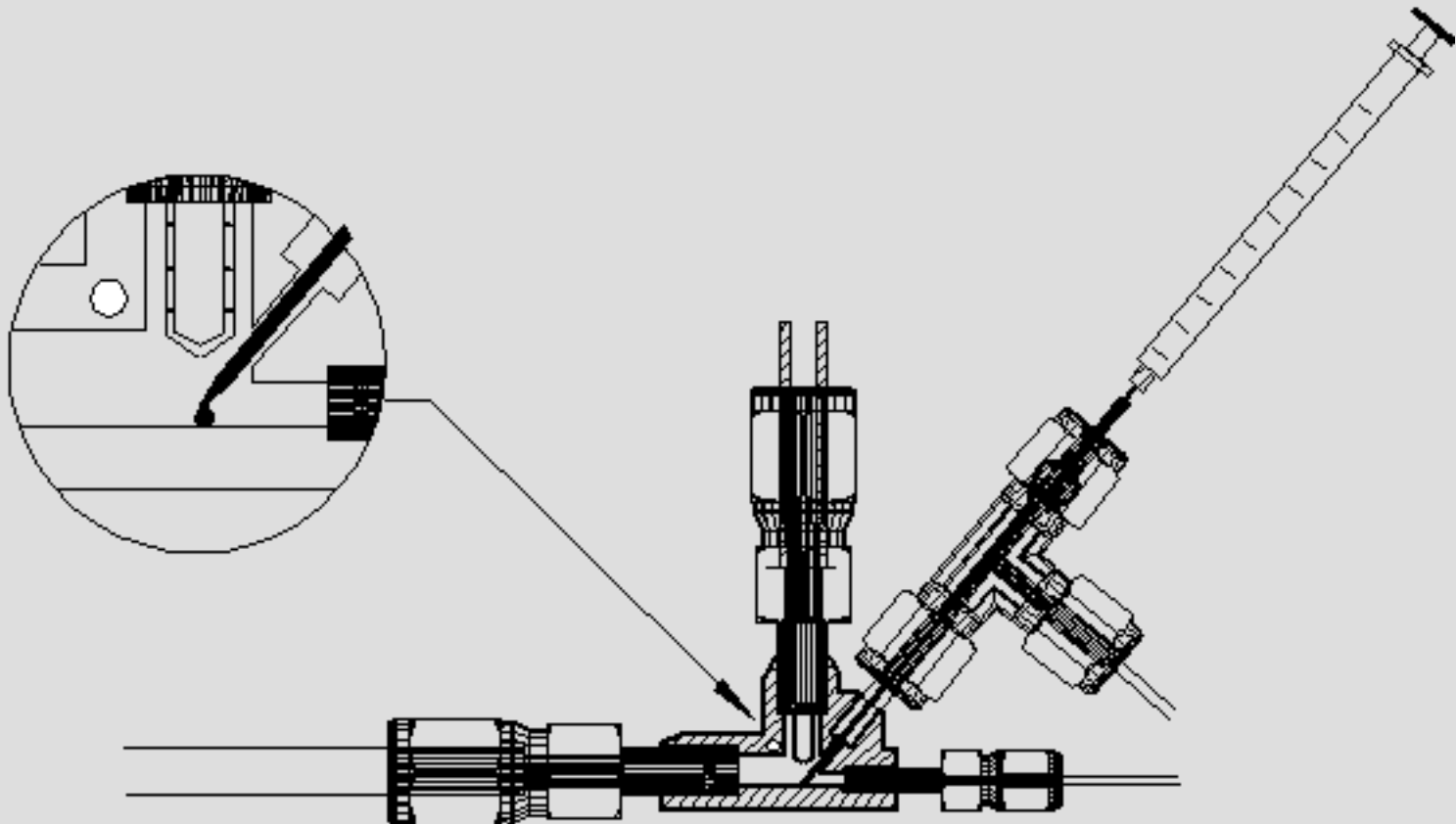
Average by
Time of Day
SOAR
Summer

TAG

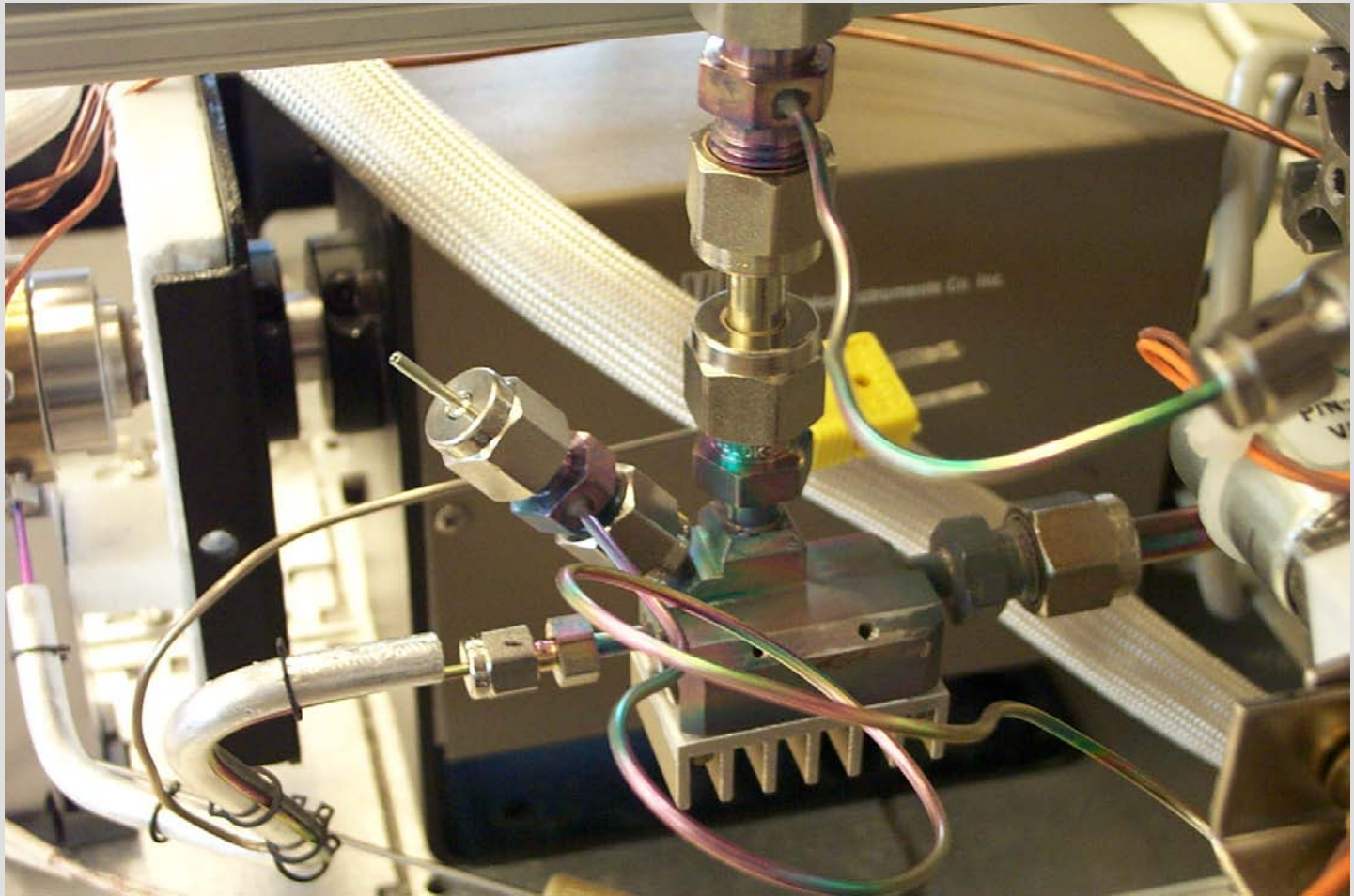


Calibrations: Liquid Standards

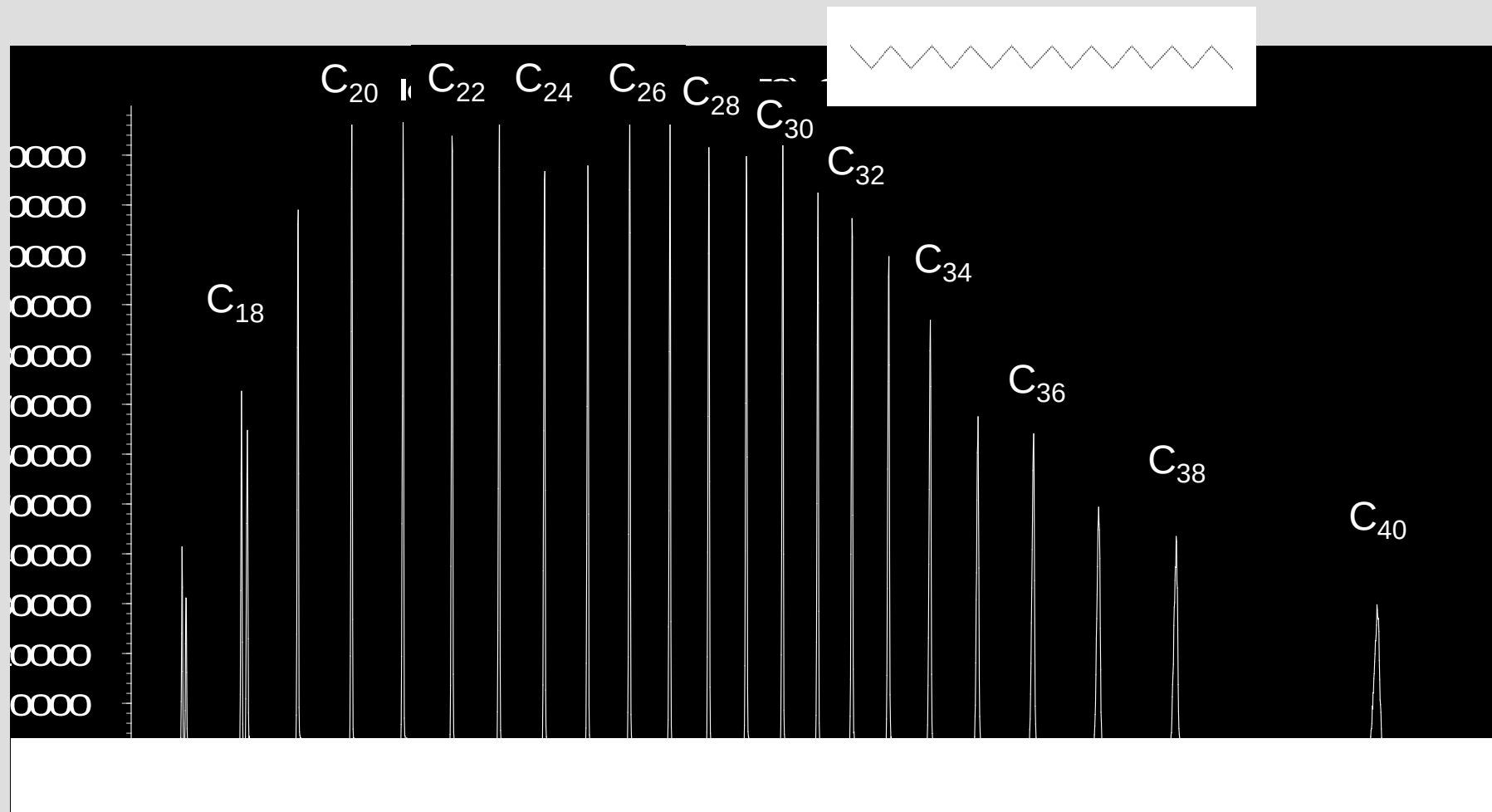
- injected directly into aerosol collection cell
- analyzed with same protocol as ambient sample



Collection Cell with Standards Injection Port



Response for n-Alkanes Standard (10ng/compound)



Balance between Standards and Time for Ambient Measurement

- **Tracking standard**

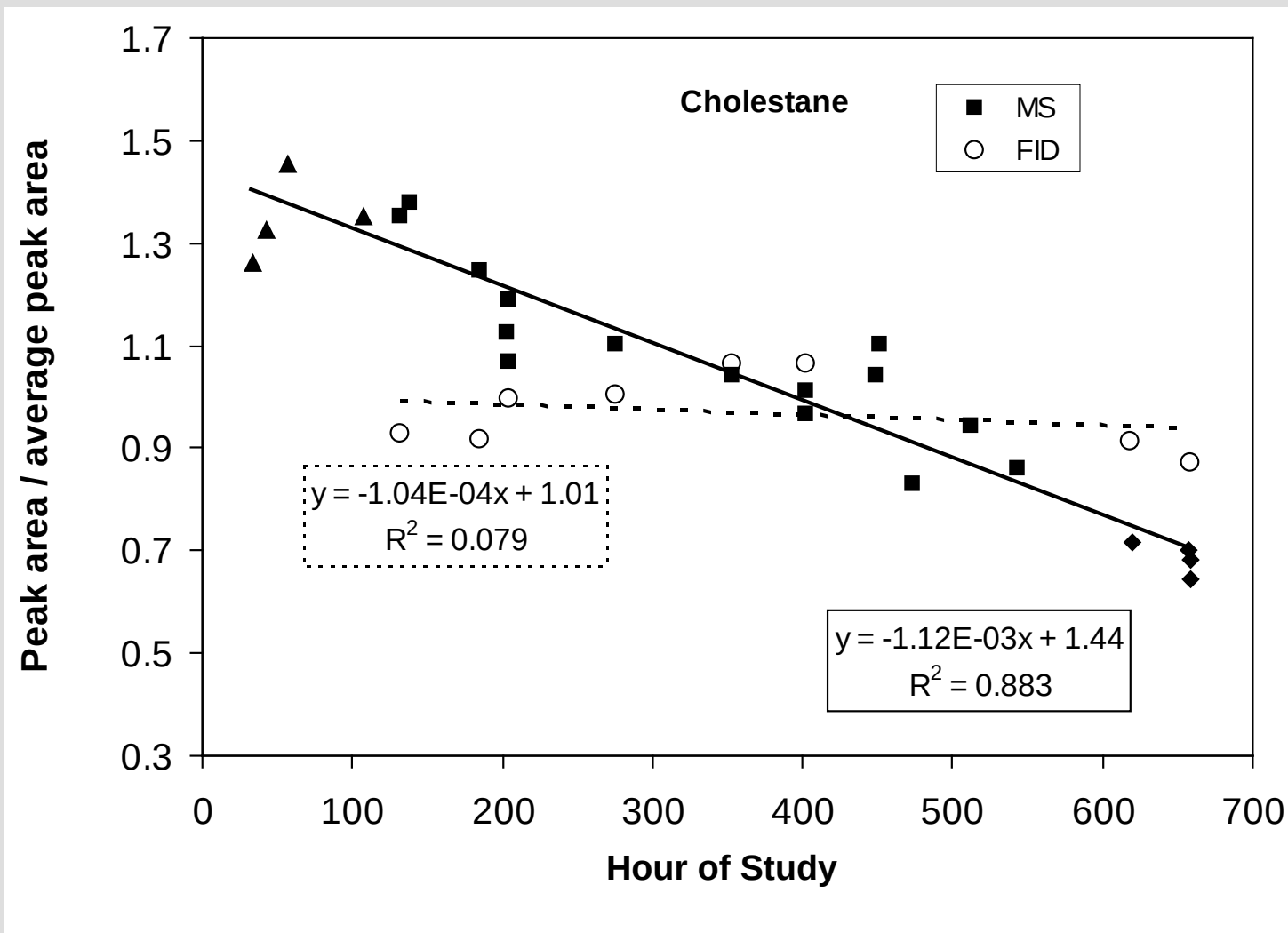
Daily @ one level (multipoint cal 1-2x/study)

- alkanes, PAHs, sterane, acids
- mix of compound classes

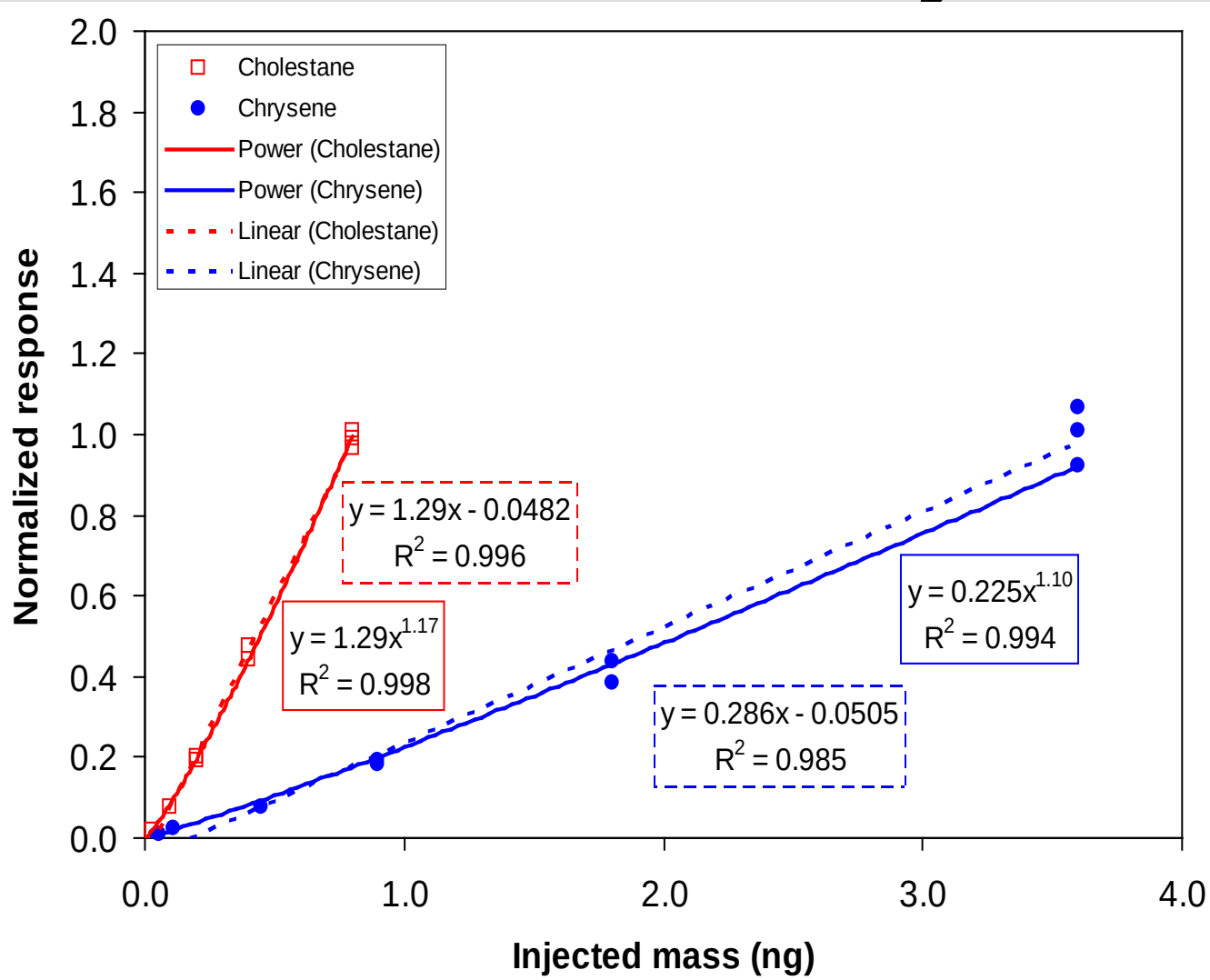
- **Authentic standards** (multipoint cals, 1-2x)

- Series of PAHs, series of alkanes, monoacids...
- Overlap of ~one compound with tracking standard

Tracking Standard Response over 25 days



Calibration Curves for Cholestane and Chrysene

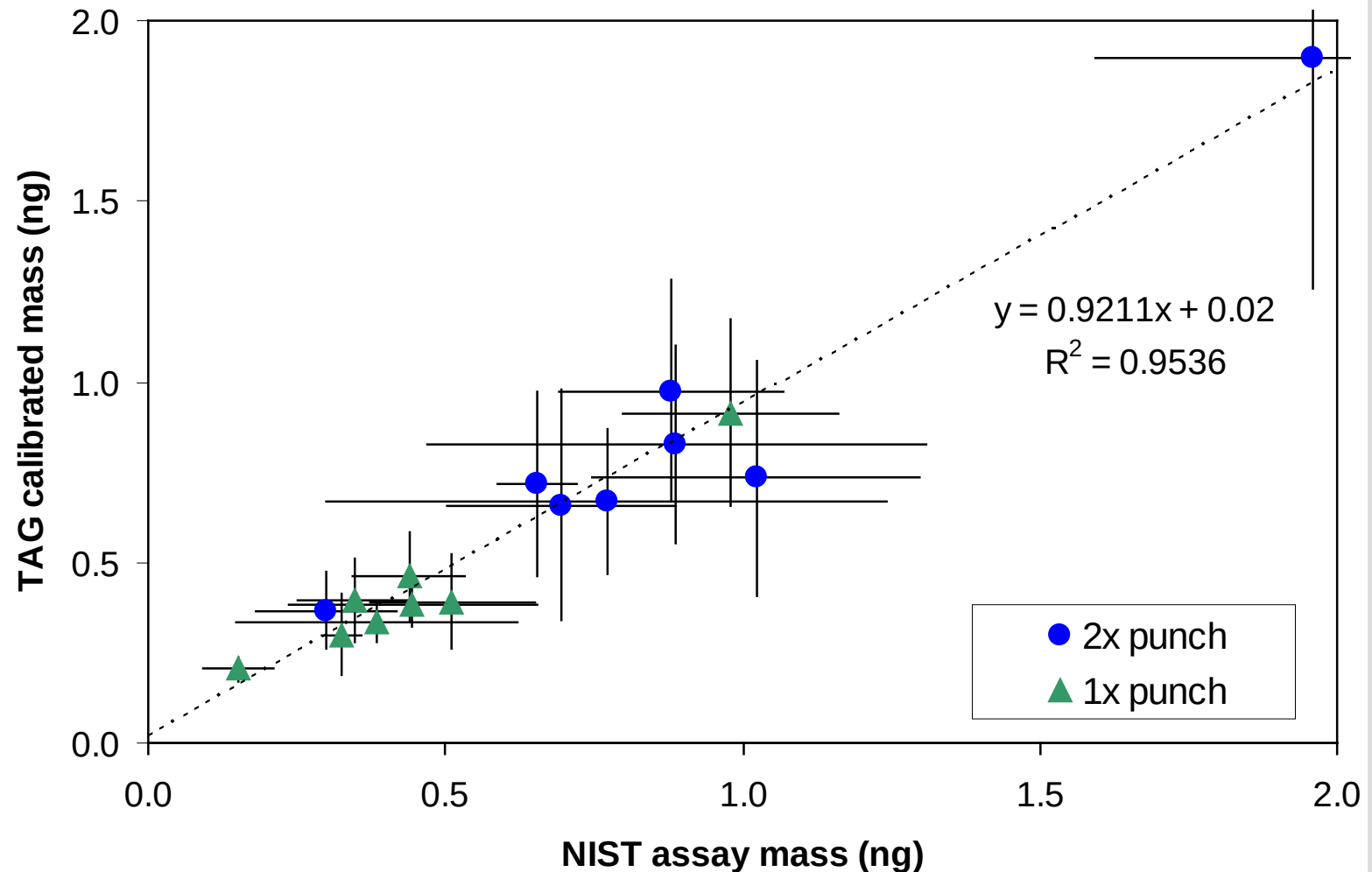


Tests with NIST Reference Filters

1 or 2 punches
inserted
directly in cell
and analyzed



TAG response Compared to NIST Multilab Assay

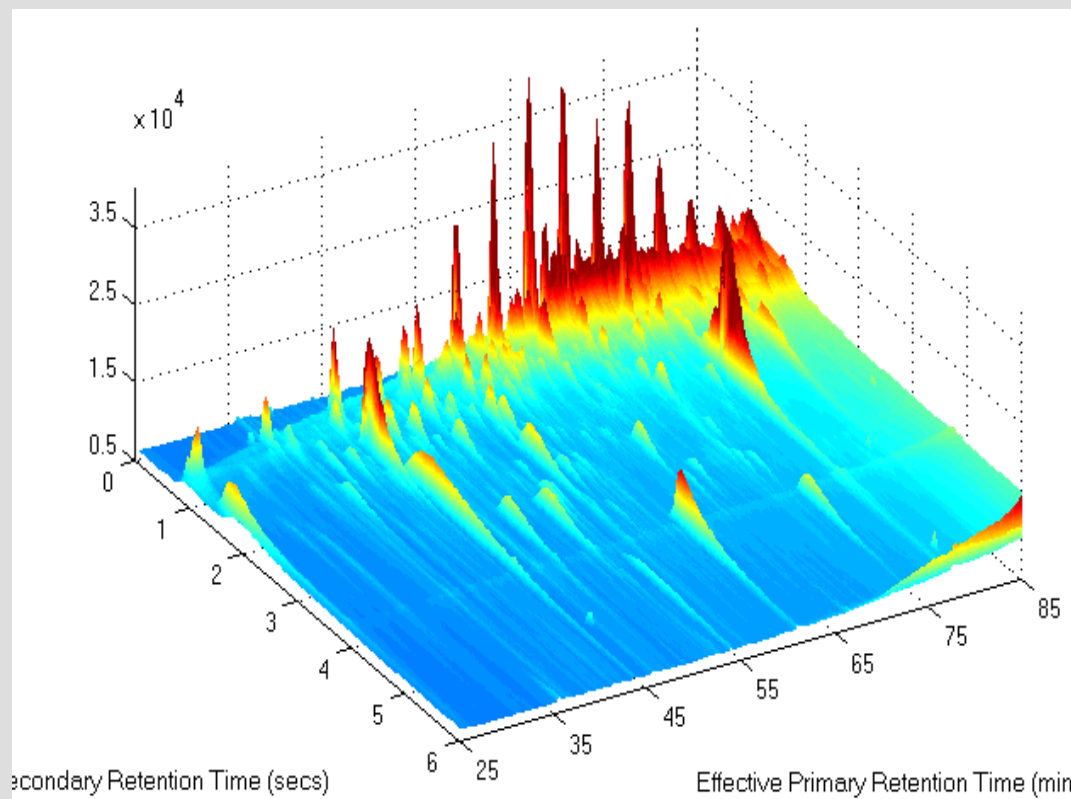


Summary: Where we have come

- Automated, In-situ instrument for identifying individual organic compounds
- Applied PMF analysis to identify sources and different types of SOA
- Calibration with liquid standards, equivalent to NIST assay for filter punches

Summary: Where we are Going

- Extending compound separation with two-dimensional chromatography
- Further improvements in Quantification



ACKNOWLEDGEMENTS

Instrument Development

DOE SBIR Phase I & II:DE-FG0203ER83825

DOE STTR Phase I & II : DE-FG0205ER86235

Field Campaigns

NOAA

California Air Resources Board